

Growing and Harvesting Nitrogenase Proteins from Azotobacter Vinelandii

Abstract

To study the worth of the electronic interaction between constituent proteins FeP and MoFeP, in the enzyme Nitrogenase, I am growing and harvesting different types of bacteria: A wild type bacteria, Azotobacter vinelandii which is naturally found and can make Nitrogenase easily, the mutant βK400E, which changes a positive ion on the MoFeP to become negative (thus causing a weaker connection between the FeP and the MoFeP), and finally L127 Δ , which takes away a part of the FeP, causing the FeP to become stuck in place to the MoFeP.

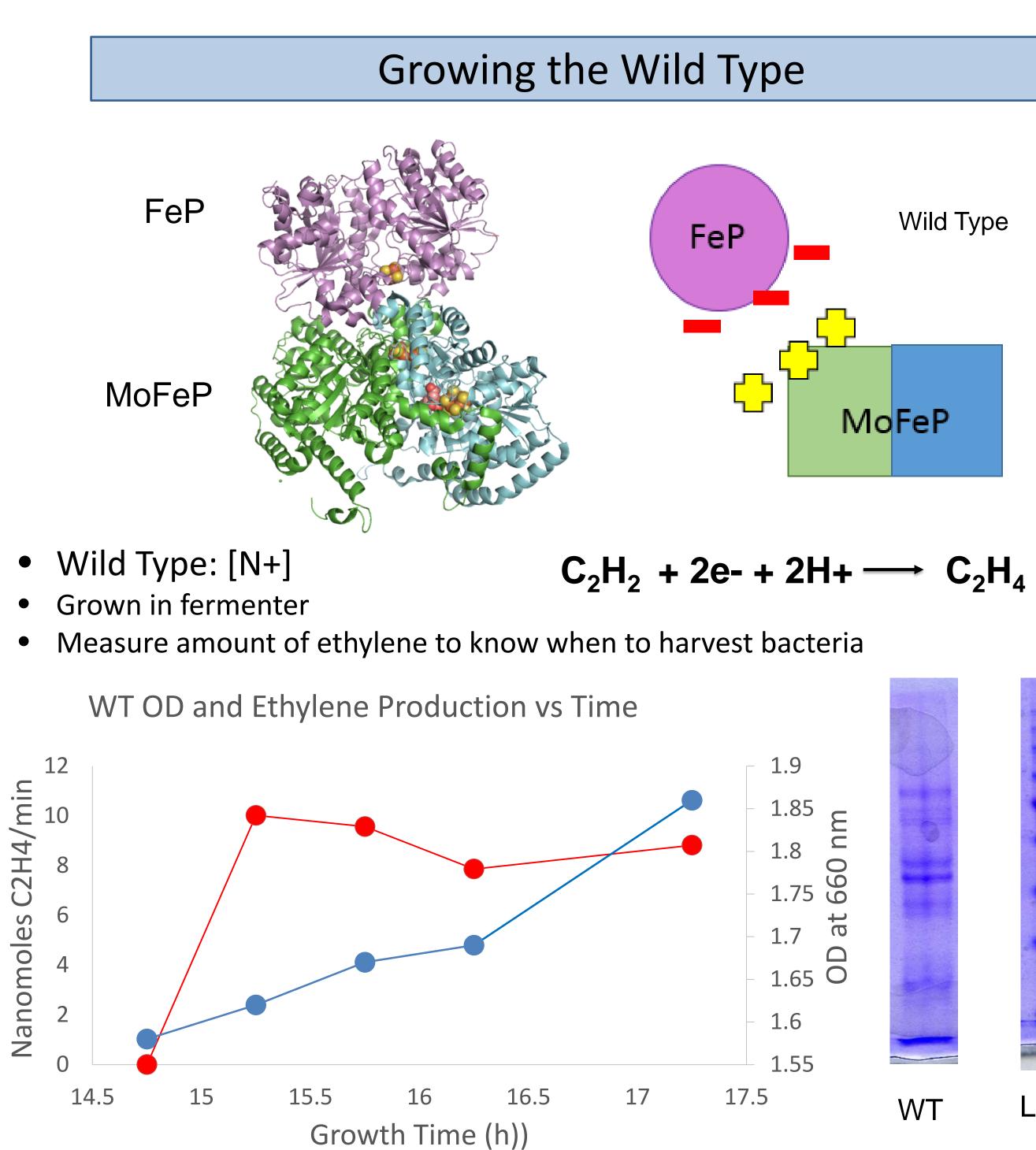
Background

Ammonia

- Needed by countless organisms
- Pharmaceuticals
- One of the highest-produced chemicals in the world
- Can be made by Nitrogen fixation (slow):

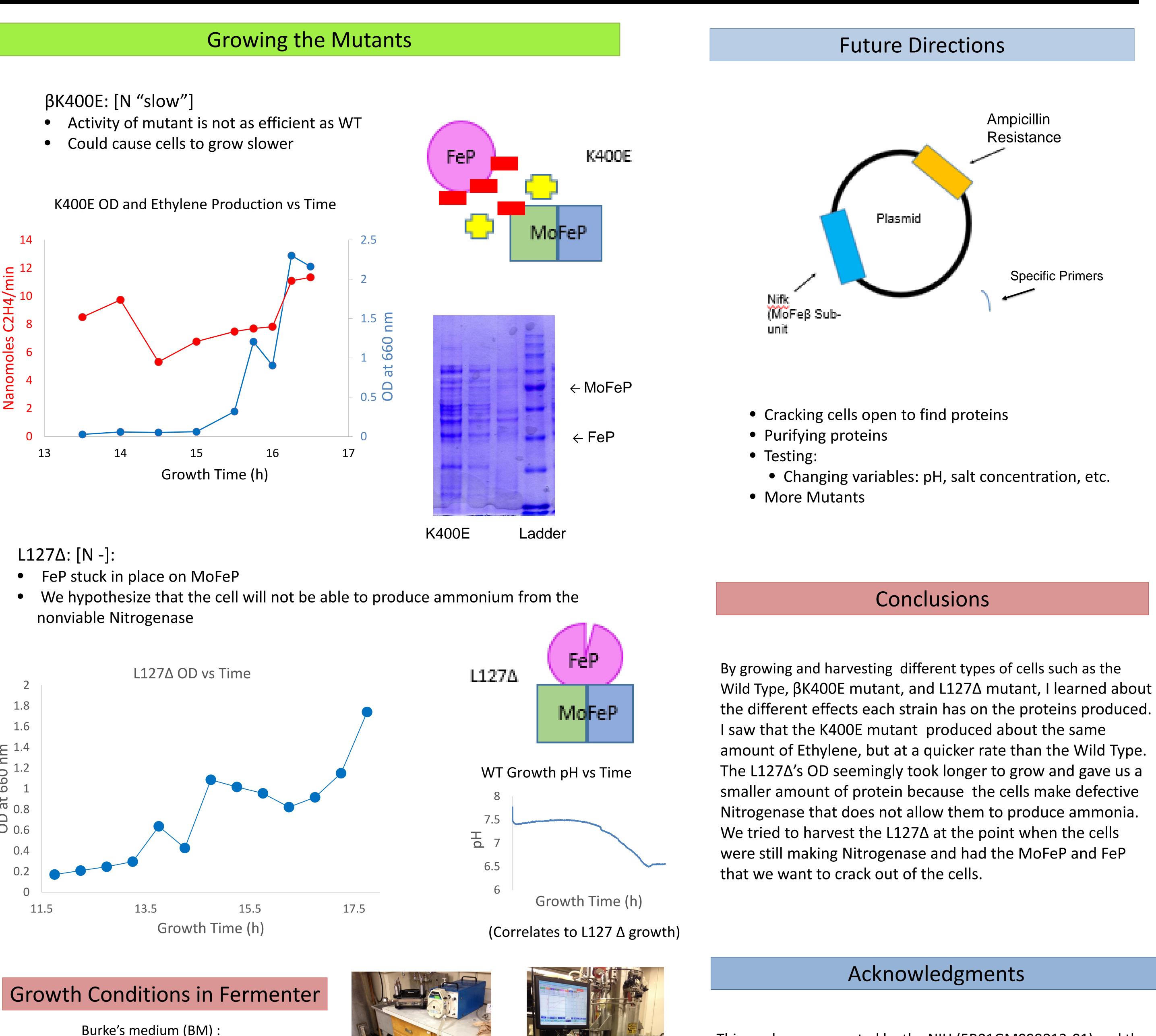
N_2 + 8 e- + 16 ATP + 8 H+ \rightarrow 2 NH₃ + 16 ADP + 16 Pi + H₂

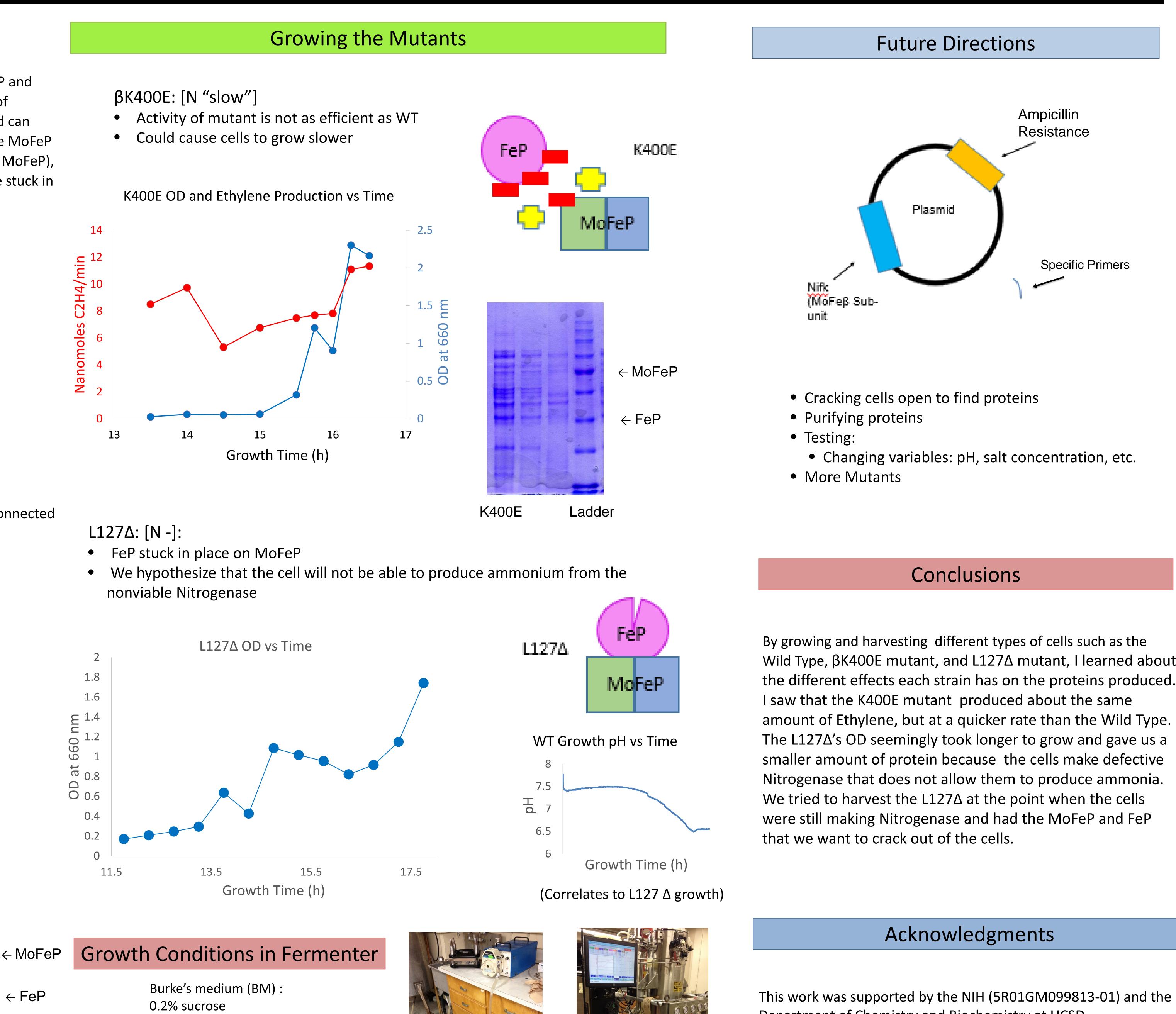
- Nitrogenase
 - Only enzyme that can turn dinitrogen from the atmosphere into ammonia
 - Within the enzyme itself, the Fe protein (FeP) transfers 1 electron to the connected MoFe protein (MoFeP) with each association



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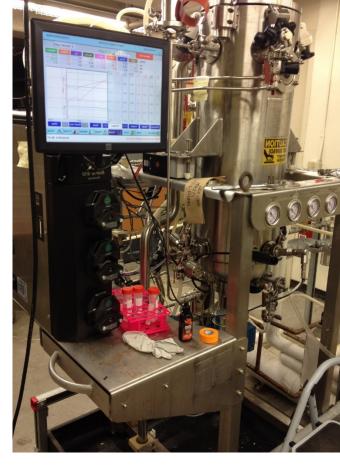


 0.9 mM CaCl_2 1.67 mM MgSO_4 $0.035 \text{ mM FeSO}_{4}$ $0.002 \text{ mM Na}_2\text{Mo}_2\text{O}_4$ 10 mM Na₃PO₄ (pH 7.4) 10 mM NH₄Cl

WT

Ladder





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